Journal of Materials Chemistry A



PAPER

View Article Online
View Journal | View Issue



Cite this: J. Mater. Chem. A, 2015, 3, 13986

Received 27th October 2014 Accepted 18th May 2015

DOI: 10.1039/c4ta05749j

www.rsc.org/MaterialsA

EM shielding effectiveness of Pd-CNT-Cu nanocomposite buckypaper†

Anil Kumar, Avanish Pratap Singh, Saroj Kumari, A. K. Srivastava, Sivaiah Bathula, S. K. Dhawan, P. K. Duttab and Ajay Dhar*

We report the synthesis of a nanocomposite consisting of Pd doped multiwall carbon nanotubes decorated with Cu nanoparticles, as a lightweight and flexible microwave absorbing material, using an electroless technique. The synthesised nanocomposite was extensively characterized by employing X-ray diffraction, Raman spectroscopy, FESEM, and HRTEM and their results were correlated with the high electromagnetic interference (EMI) shielding observed in the present study. The optimum dielectric properties coupled with good electrical conductivity of this nanocomposite contribute to designing this absorption-based microwave shield, which exhibited a good EMI shielding effectiveness (EMI SE) of \sim 35 dB at a thickness of 200 μ m, resulting in a high specific EMI SE of \sim 108 dB cm³ g⁻¹ in the Ku-band.

1. Introduction

Electromagnetic interference (EMI) has received worldwide social and scientific attention due to the extensive use of electronic and communication gadgets.¹⁻³ Serious problems of EMI are generally encountered in satellite communication, radar surveillance systems, digital devices, wireless technology and other sophisticated electronic equipment.^{1,3} Besides this, EMI in daily life is also known to incur harmful effects under certain environments.4 This novel sort of pollution has triggered the use of efficient counter-measures to ensure the optimum performance of electronic gadgets in the presence of EM noise,5 especially for aircraft and aerospace applications. As lightweight and flexibility are essential considerations, especially for aircraft and aerospace applications, enormous efforts are presently underway for developing a suitable EMI shielding material on an industrial scale.6 Light weight carbon-based polymer composites have been reported to possess high EMI shielding along with excellent mechanical properties with large flexibility in their aspect ratio.7 Several such studies have been reported employing different polymer reinforced carbon composites8 but they suffer from the inherent problems of low melting temperature and low temperature degradation characteristics,9 which make them unsuitable for most of the applications.

Composites, based on different forms of highly conducting carbon, such as carbon nanotubes, ¹⁰ graphene, ¹¹ and carbon foam, ^{12,13} have been reported to achieve high EMI shielding and the shielding effectiveness (SE) of these polymer composites is found to decrease as per the following order: multiwall carbon nanotubes (MWCNTs) > graphene > carbon foam > carbon nano-fibers > carbon black. ¹⁴ The EMI SE of the polyaniline composite with Ag decorated (5.0 wt% loading) graphene was reported to be 29.33 dB, ¹⁵ which was attributed to the enhancement in electrical conductivity due to metal particle decoration and high permittivity of these composites, and it was observed that absorption was the governing factor for the improved EMI shielding. Shinn-Shuong *et al.* ¹⁶ have reported an EMI shielding of ~30 dB in Cu/Ni coated fiber reinforced composites, developed by an electroless technique.

In the same order, CNT based composites, apart from being lightweight, exhibit good electrical conductivity and mechanical flexibility with high EMI shielding.17 However, the decoration of metal particles on these nano-sized matrices could further expand their potential as excellent EMI shielding materials due to the increase in their charge carrier density.18 Such metal nanoparticle decorated composites could provide distinctive advantages over conventional EMI shielding materials as they exploit the combined properties of both the components, i.e., excellent electrical and thermal conductivities of the metal with a low thermal expansion coefficient and corrosion-resistive properties of the CNTs.19 Thus, metal nanoparticle decoration on MWCNTs could be a novel and an economical idea for EMI shielding applications. In this direction, we have previously reported20 a fabrication method for high strength carboncopper (C-Cu) composites from a polyaromatic hydrocarbon based carbon copper nanocomposite synthesized using an electroless technique, which possess a very good EMI shielding

^aMaterial Physics and Engineering Division, CSIR-National Physical Laboratory, Dr K. S. Krishnan Marg, New Delhi-110012, India. E-mail: adhar@nplindia.org; Fax: +91-11-25726938; Tel: +91-11-45609456

^bDepartment of Chemistry, MN National Institute of Technology, Allahabad-211 004, India

[†] Electronic supplementary information (ESI) available. See DOI: 10.1039/c4ta05749j

of \sim 58 dB with a low reflection of \sim 12.5 dB at a sample thickness of 2 mm. Some techniques that have been so far employed for the decoration of metal nanoparticles on MWCNTs include impregnation, electroless metal-plating, self-assembly, electrodeposition, hydrological vapor deposition, etc. Despite the good EMI shielding characteristics of highly conducting materials, such as carbon nanotubes and other exotic forms of carbon, only a few attempts have been made to introduce metals along with MWCNTs for enhancing the shielding effectiveness. However, W. Zhao et al. reported silver nanoparticle decorated CNTs by an *in situ* photochemical reduction method with a good EMI shielding. However, due to their cost they may not be a commercial material.

In the present study, we demonstrate the synthesis of a costeffective Cu nanoparticle decorated Pd doped MWCNT derived C-Cu nanocomposite buckypaper²⁹ (thickness = $200 \mu m$) as an efficient EMI shielding material employing an electroless technique. This technique, based on simple chemical reduction chemistry, is a fast and reproducible synthesis technique, wherein the synthesis of Cu nanoparticles and their decoration on MWCNTs take place in situ. In this process, liquid ammonia and hydrazine hydrate are used as the complexing and reducing agents,30 and Pd may be used as a mediator for interaction between MWCNTs and Cu nanoparticles.31 The transmission electron microscopy results indicated a uniform distribution of Cu nanoparticles (2-5 nm) on the CNTs (diameters between 20 and 60 nm). The selected area electron diffraction (SAED) pattern confirms the presence of Cu particles with 111, 200 and 220 crystal planes. The Cu nanoparticle decorated Pd doped MWCNT buckypaper, having different Cu nanoparticle loadings, is found to possess a very low density (\sim 0.25 to 0.40 g cm⁻³). The synthesized buckypaper having a thickness of 200 μm at 15% Cu loading are found to exhibit an EMI SE of \sim 35 dB (specific EMI SE ${\sim}108~\text{dB cm}^3~\text{g}^{-1}$), which, coupled with their good flexibility32 and lightweight could make them excellent candidates for large-scale EMI shielding applications.

2. Experimental

2.1 Material synthesis technique

MWCNTs, used in the present study, were procured from Nonosil and palladium chloride (99.99%), copper sulphate (CuSO₄·5H₂O) (99%), liquid ammonia, and hydrazine hydrate were procured from Merck, India. Initially, the purified MWCNTs (1 g) are taken in 100 mL of ethyl alcohol and sonicated for \sim 15 minutes to obtain dispersed MWCNTs. An acidic solution (5 mL HCl mixed in 500 mL of de-ionized (DI) water) having palladium chloride (PdCl₂, 0.02 g L^{-1}) and a small amount of SnCl₂ (0.001 g L⁻¹) was then added to these dispersed MWCNTs for Pd doping. The mixture was then sonicated continuously for another 30 minutes and finally filtered. A brief description of the activation reaction has been discussed in the ESI (eqn I and II†). The filtrate was dried at 300 °C with a heating rate of 100 °C min⁻¹, which results in Pd doped MWCNTs, denoted as PdCNTs in the subsequent discussions. The PdCNT powder was coated with Cu nanocrystalline particles with different Cu loadings (5, 15 and 25 wt%) by an electroless technique. 31 First, for the coating of Cu nanoparticles, bath solutions having 1 L of DI water and containing an appropriate amount of Cu ions, derived from the reduction of cupric sulphate (CuSO₄·5H₂O) for various Cu loading concentrations (5, 15 and 25 wt%), were separately prepared. In the bath solution, liquid ammonia was used as a complexing agent and the bath solutions were subjected to constant stirring throughout the experiment. In order to obtain uniform coating, the pH and temperature of bath solutions were maintained within 8-9 and 40-50 °C, respectively, using liquid ammonia solution. At the optimized temperature and pH, hydrazine hydrate (10 mL) was added as a reducing agent to form Cu nanoparticles. This reaction was continued for \sim 1 h to form Cu nanoparticle decorated PdCNT nanocomposites with varying Cu loadings on PdCNTs (as shown in Fig. 1). These nanocomposites will be denoted as Cu_x PdCNTs, where x = 0, 5,

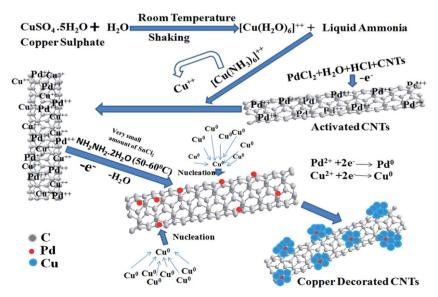


Fig. 1 Schematic of the synthesis mechanism for the decoration of Cu particles on PdCNTs.

15 and 25 represent the wt% loading of Cu nanoparticles on PdCNTs. The PdCNTs and nanocomposite network sheets (buckypaper) were fabricated using a vacuum filtration method³³ for electrical and EMI shielding measurements. A filter paper (Whatman no. 44) was used for filtration and porous metal sheets were used for the preparation of buckypaper of CNTs under rotary vacuum. The dispersed decorated CNTs were slowly poured onto the flat wet filter paper and after filtration they were allowed to dry. Finally, the buckypaper was dried for several hours at 80 °C. This oven dried buckypaper was sintered at 500 °C in a nitrogen atmosphere.

2.2 Material characterisation techniques

Field emission scanning electron microscopy (FESEM) (ZEISS-EVO MA10) and transmission electron microscopy (TEM) (FEI Tecnai T30) were used to study the morphology and microstructural characteristics of the synthesized sample. The physical and chemical nature of the composite was analyzed using a Raman spectroscope (Renishaw inVia, Macro Raman) having a 514.5 nm laser for excitation. The electrical conductivity was determined using a Keithley programmable current source (Model-6221) and a nano-voltmeter (Model-2182A). The buckypaper cut in rectangular sheets ($13 \times 7 \times 0.2 \text{ mm}^3$) and having four ohmic contacts on each end, made using silver paste, has been used for electrical measurements. The phase identification and crystallite size estimation were carried out using X-ray powder diffraction (XRD) (Rigaku, MiniFlexII). To explore the EMI-SE and dielectric measurements of the nanocomposites, a Vector Network analyzer (model-E8362B) was employed on rectangular shaped (15.8 \times 7.9 \times 0.2 mm³) buckypaper inserted in a Cu sample holder connected between the wave-guide flanges of the network analyzer for operation in the Ku-band (12.4-18.0 GHz).

3. Results and discussion

The XRD pattern of the $\text{Cu}_{25}\text{PdCNT}$ nanocomposite is shown in Fig. 2(A). The peak observed at $2\theta=26.1^\circ$ corresponds to the hexagonal (002) plane of MWCNTs and the smaller peaks at $2\theta=43.8^\circ$ and 53.4° represent (111) and (200) fcc planes of Cu, ³⁴ as indicated by the enlarged image of the XRD pattern (inset of

Fig. 2(A)), which was also confirmed by SAED patterns. The Raman spectra (Fig. 2(B)) of MWCNT, Cu₀PdCNT and Cu₂₅-PdCNT nanocomposites show two clear bands at 1577 cm⁻¹ (G band) and 1359 cm⁻¹ (D band), which confirm the presence of carbon. The Raman spectra also indicate the disorder induced (D-band) and graphitized carbon (G-band) 2D bands that appear at 2709 cm⁻¹. Here, we have observed a change in the intensity ratio of $I_{\rm G}/I_{\rm D}$, which may be due to metal doping in the MWCNT structure as unlocalized free electrons from Cu create an effective change in E_{2g} photon corresponding to sp² atoms (G band). The same structure as unlocalized free electrons from Cu create an effective change in E_{2g} photon corresponding to sp² atoms (G band).

The microstructure of Cu particle decorated PdCNTs (Cu = 5, 15 and 25 wt% loading) is shown in Fig. 3(A-D). Fig. 3(A) depicts a decoration of Cu nanoparticles on PdCNTs with 5% Cu loading and suggests that the diameter of MWCNTs is \sim 20-60 nm. Fig 3(B) shows the image of the 15 wt% Cu loaded nanocomposite, which suggests a reasonably uniform distribution of Cu nanoparticles in the nanocomposite. However, on further increasing the Cu wt% in the nanocomposite (>15%), the agglomeration is clearly evident as shown in Fig. 3(D). Elemental analysis of the nanocomposite is shown in the inset of Fig. 3(D), which confirmed the presence of carbon and Cu in the nanocomposite. Palladium has been confirmed in PdCNTs by the elemental analysis to be very small in quantity and the corresponding details are given in Fig. 1S(A-C).† However, Pd is not detected in the case of the CuxPdCNT nanocomposite, which is due to the small quantity (0.005 g in the 1 g CNTs) of Pd used during the synthesis. The corresponding quantitative elemental analysis and elemental distribution mapping on Cu₁₅PdCNTs are shown in the ESI (Fig. 2S(A-C)†). Thermogravimetric analysis of the Cu₁₅PdCNT nanocomposite also was conducted in air and inert atmospheres for the confirmation of wt% Cu present and this result confirmed that \sim 15% Cu was present in the nanocomposite, Fig. 2S(D).† The microstructure of Cu₁₅PdCNT buckypaper exhibits a uniform decoration of Cu nanoparticles on PdCNTs, which are entangled with each other and for detail description see Fig. 3S (as ESI†).

Transmission electron microscopy (TEM) experiments were conducted on the Cu_xPdCNTs samples to show the decoration of Cu nanoparticles in CNTs. It was noted that the Cu nanoparticles of 2 to 5 nm in size (marked with arrows) were distributed on a CNT of a diameter of about 35 nm in

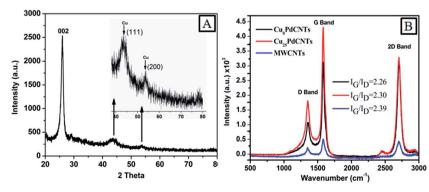


Fig. 2 (A) XRD pattern of the $Cu_{25}PdCNT$ nanocomposite; (B) comparative Raman spectra of $Cu_0PdCNTs$ and $Cu_{25}PdCNTs$ nanocomposites.

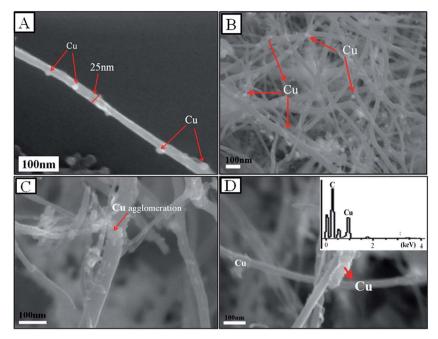


Fig. 3 FESEM image of Cu_x PdCNT nanocomposites (A) Cu_5 PdCNT nanocomposite; (B) Cu_{15} PdCNT buckypaper; (C) and (D) Cu_{25} PdCNT nanocomposite; the inset in (D) shows the EDX analysis of the Cu_{25} PdCNTs.

Cu₅PdCNTs (Fig. 4S(A)†). The low resolution TEM image of Cu₁₅PdCNTs (Fig. 4(A)) displayed a dense-distribution of multiwalled carbon nanotubes (MWCNTs) in general. However, the magnified image revealed a uniform distribution of Cu NPs (marked with a set of arrows) on MWCNTs (Fig. 4(B)). It was interesting to note that the presence of Cu was either on the outer surfaces of the carbon nanotubes or trapped in the multiwalled structures of individual nanotubes. The inset in Fig. 4(B) shows a Cu nanoparticle present in the core region of an individual carbon nanotube with a diameter of about 15 nm and length covering approximately 40 nm. Fig. 4(C) shows an ultrafine Cu nanoparticle mostly on the surface of a nanotube of about 40 nm in outer diameter of the tube. Atomic scale imaging results show that the interlayer separation of the multiwall is approximately 0.34 nm (inset in Fig. 4(C)). Although, in general, the interlayer separation of the walls is quite parallel, the regions around the nanoparticle (approximately 10 nm) are disturbed with the wavy geometry of interlayers of nanotubes (Fig. 4(D)). It is noted explicitly that the Cu nanoparticles are intact and very cohesive with the surfaces of the nanotube. The gray level contrast and high resolution images were obvious to distinguish the carbon nanotubes associated with Cu nanoparticles in the composite microstructure. However, further revelation (4S(B)†) was possible by recording the atomic scale image of Cu and subsequent identification of the 0.21 nm interplanar spacing of Cu, along 111 planes (crystal structure: fcc, lattice, ref: JCPDS copper file no. 04-0836). Moreover, a corresponding fast Fourier transform (FFT) of the composite micrograph delineates the presence of 0.34 nm and 0.21 nm spacing that evolved from MWCNTs and Cu, respectively, in the reciprocal space (inset in 4S(B)†). A micrograph of the Cu₂₅PdCNT nanocomposite is shown in

Fig. 4S(C),† which clearly exhibits the agglomeration of Cu nanoparticles on further increasing the Cu% loading.

The room temperature DC electrical conductivity of Cu_xPdCNT nanocomposites, investigated by a standard four-

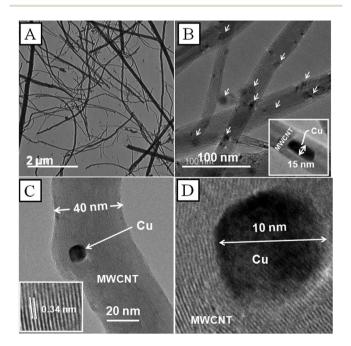


Fig. 4 TEM micrographs of Cu $_{15}$ PdCNTs showing the composite microstructure of MWCNTs and Cu NPs: (A) large area distribution of MWCNTs and Cu, (B) high magnification micrograph of nanotubes with nanoparticles, (C) single nanoparticle on the outer surface of a CNT, and (D) cohesive microstructure of the multiwall of a CNT with a nanoparticle. Insets: nanoparticle inside the CNT (B), resolution of the multiwall (C).

probe technique, is shown in Fig. 5. Since both Cu and MWCNTs possess very good electrical conductivity,26 the resultant electrical conductivity of CuxPdCNT nanocomposites escalates sharply with increasing wt% loading of Cu i.e. from 3.64 S cm^{-1} (Cu₀PdCNTs) to 11.64 S cm^{-1} (Cu₁₅PdCNTs). Interestingly, Fig. 5 suggests that further increasing the loading of Cu nanoparticles to 25 wt% shows an adverse effect on the electrical conductivity and it decreases to 8.95 S cm⁻¹, which suggests that the optimum threshold loading limit of Cu nanoparticles is around 15 wt%. This may be due to the agglomeration of Cu nanoparticles with Cu loading >15 wt%, which is in agreement with the TEM results on the Cu25PdCNT nanocomposite (Fig. 4(D)). A similar phenomenon has also been reported earlier in Ag nanoparticle decorated CNTs, which may lead to a decrease in the conductivity and other properties

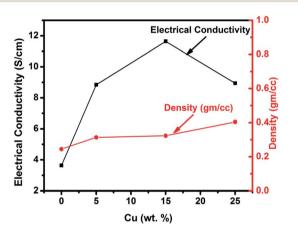


Fig. 5 Variation of DC electrical conductivity and density with Cu wt% nanoparticle loading in $Cu_x PdCNT$ nanocomposites (where x = 0, 5, 15and 25 wt%).

of the nanocomposites.³⁷ This suggests that a Cu loading ≤15 wt% results in a uniform distribution of Cu nanoparticles on PdCNTs, leading to an enhanced electrical conductivity in Cu nanoparticle decorated PdCNT nanocomposites. 18 The density variation of the Cu_xPdCNT nanocomposite as a function of wt% of the Cu nanoparticle loading in PdCNTs is also shown in Fig. 5. This figure suggests that the density of the Cu_xPdCNT nanocomposite continues to increase with increasing Cu nanoparticle loading and increases from 0.24 g cm⁻³ (unloaded) to 0.40 g cm⁻³ (25 wt% Cu particle loading). This increase in the density of the Cu_rPdCNT nanocomposite with higher Cu nanoparticle loading is attributed to the high density of Cu compared to MWCNTs.

The relative complex dielectric parameters have been estimated from experimental scattering parameters $(S_{11} \& S_{21})$ obtained by standard Nicolson-Ross and Weir theoretical calculations. 38 The estimated dielectric constant (ε') symbolizes the amount of polarization occurring in the material or the storage ability of the electrical energy, while the dielectric loss (ε'') signifies the dissipated electrical energy. The frequency dependence of ε' and ε'' in the Ku band (12.4 to 18 GHz) for all the CuxPdCNT nanocomposites and buckypaper is plotted in Fig. 7(a and b). The effective value of real and imaginary parts of permittivity, for the entire frequency range, was extracted by averaging over 201 data points. The overall averaged values of ε' are found to be \sim 20.10, 41.01, 47.14 and 19.56, while that of ε'' are ~7.83, 23.25, 29.63 and 10.88 for 0, 5, 15 and 25 Cu wt%, respectively. An increasing trend in these values was observed for a Cu nanoparticle loading up to 15 wt%, which falls drastically on further increasing the Cu nanoparticle loading to 25%, which is nearly to the conductivity pattern as shown in see Fig. 5, Hence, it can be concluded that increasing the wt% loading of Cu nanoparticles (upto 15%) shows enhancement in the values of ε' and ε'' , which may be due to the higher

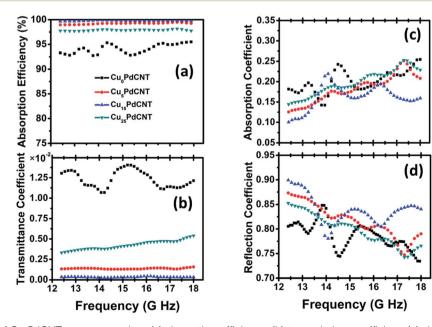


Fig. 6 EMI parameters of Cu_xPdCNT nanocomposites, (a) absorption efficiency, (b) transmission coefficient, (c) absorption coefficient and (d) reflection coefficient.

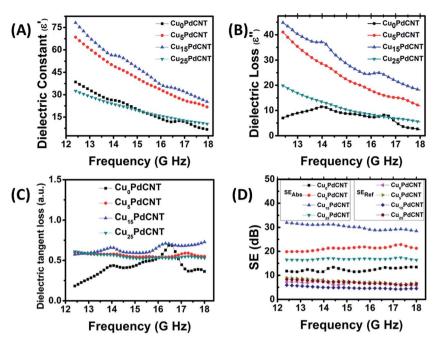


Fig. 7 Frequency dependence of complex permittivity of Cu_xPdCNT nanocomposites (A) real part (ε'); (B) imaginary part (ε''); (C) dielectric tangent loss and (D) EMI SE due to absorption (SE_{Abs}) and reflection (SE_{Ref}).

conductivity of Cu_xPdCNT nanocomposites compared to $Cu_{25}PdCNTs$. Further, the resulting relative complex permittivity, which is a measure of the polarizability of a material, induces dipolar and electric polarization in the presence of microwaves.

According to EM theory, dielectric losses are the result of complex phenomena like natural resonance, dipole relaxation and electronic polarization. In the present nanocomposites, Pd may act as a polarized centre on the surface of the MWCNTs. The high aspect ratio of MWCNTs decorated with Cu nanoparticles also enhances the absorption properties. The existence of interfaces between Cu nanoparticles and MWCNTs is also responsible for interfacial polarization, which further contributes to dielectric losses. Interfacial polarization occurs in heterogeneous media due to the accumulation of charges at the interfaces. The increase in the averaged values of ε'' with the higher wt% loading of Cu nanoparticles (≤15%) is attributed to the formation of a connecting network between PdCNTs and Cu nanoparticles in Cu_xPdCNT nanocomposites. In addition, the conducting nature of MWCNTs is further helpful in enhancing the dielectric loss.26 The natural resonances in the Ku-band can be attributed to the small size of Cu nanoparticles on the PdCNT matrix as the anisotropy energy of fine-size materials, especially in the nanoscale, 21,25 is expected to be higher due to the surface anisotropic field, termed as the size effect,4,39 which contributes to the enhancement of the microwave absorption.

Furthermore, the reflection coefficient (R), transmittance coefficient (T), absorption coefficient (A), and absorption efficiency of the nanocomposites were evaluated using S parameters as, $R = |S_{11}|^2 = |S_{22}|^2$, $T = |S_{21}|^2 = |S_{12}|^2$, A = 1 - R - T and

 $A_{\rm eff} = (1-R-T)/(1-R)$, respectively. Fig. 6(a-d) show the calculated absorption efficiency, T, A and R. The T value of $Cu_xPdCNTs$ is found to be minimum for $Cu_{15}PdCNT$ (less than 0.1) *i.e.* more EM waves are consumed by the $Cu_{15}PdCNT$ nanocomposite, which leads to a more significant decrease of the T values, as shown in Fig. 6(b). As shown in Fig. 6(c), the A values of the Cu_xPdCNT nanocomposites were in the range of 0.1–0.25. The R values of the Cu_xPdCNT nanocomposites were in the range of 0.75–0.90. Moreover, the absorption efficiency (%) of the Cu_xPdCNT nanocomposites increases with increasing weight% of Cu (Fig. 6(a)) and $Cu_{15}PdCNTs$ show a % absorption efficiency more than 99.9%.

When EM radiation is incident on a shield, some part of the incident wave is absorbed and some part is reflected from the surface and the rest is transmitted through the shield. Hence, the total EMI SE (SE $_{Tot}$) is the sum of the contributions from absorption (SE $_{Abs}$), reflection (SE $_{Ref}$), and transmission or multiple reflections (SE $_{Mlt}$). Therefore the EMI SE of any material can be expressed as.

$$SE_{Tot} (dB) = SE_{Ref} + SE_{Abs} + SE_{Mlt} = 10 \log(P_T/P_I)$$

= 20 \log(E_T/E_I) (1)

where, $P_{\rm I}\left(E_{\rm I}\right)$ and $P_{\rm T}\left(E_{\rm T}\right)$ are the incident and transmitted power (electric field intensity) of EM waves, respectively. The correction term ${\rm SE_{Mlt}}$ can be ignored for practical applications when ${\rm SE_{Tot}} > 10$ dB. 46,47 Hence, the equation (eqn (1)) reduces to:

$$SE_{Tot} (dB) \approx SE_{Ref} + SE_{Abs}$$
 (2)

Furthermore, SE_A and SE_R are obtained by using following equations:

$$\begin{aligned} \mathrm{SE}_{\mathrm{Abs}}(\mathrm{dB}) &= 20\{t/\delta\}\log e = 20d\sqrt{\mu\omega\sigma_{\mathrm{ac}}/2}\log e = 8.68\{t/\delta\} \\ &= 8.68t\sqrt{\sigma\omega\mu'/2} \end{aligned}$$

(3

$$SE_{Ref} (dB) = 10 \log\{\sigma_{ac}/16\omega\varepsilon_0\mu'\}$$
 (4)

The σ_{ac} and δ can be related to the imaginary permittivity (ε'') and real permeability (μ') as $\sigma_{ac} = \omega \varepsilon_0 \varepsilon''$ and $\delta = \sqrt{2/\sigma \omega \mu'}$, respectively.

The above equations suggest that electrical conductivity and complex permittivity are two governing parameters for the suppression of reflection or improvement in microwave absorption. Therefore, the decoration of Cu nanoparticles (more conducting) in PdCNTs (lesser conducting) in an appropriate amount is expected to improve the absorption of EM waves. The variations of SEAbs and SERef over the frequency range of 12.4–18 GHz are shown in Fig. 7(D), from which it can be concluded that the SE due to SEAbs for CuxPdCNT nanocomposites, when averaged over the entire frequency range, has been found to vary from 12.36 to 30.16 dB but SE_{Ref} has very less variation of 6.5 \pm 1.7 dB. Furthermore, the sample Cu₁₅PdCNTs show a maximum SE_{Tot} of 34.98 dB which is conquered by SE_{Abs} (30.15 dB) and partially shared by SE_{Ref} (4.82 dB). The SE_{Tot} achieved for the Cu decorated nanocomposites is significantly higher than that of the Cu₀PdCNTs, which shows that Cu decoration plays a crucial role in the effective enhancement of SE. The SE_{Abs} becomes more dominant as compared to the SE_{Ref} in the microwave range.44 From the detailed shielding analysis, it has been observed that the SE_{Tot} value increases with an increase in the Cu content and when the Cu content reaches 15 wt% in PdCNTs, the observed SE_{Ref} value <5 dB, which is the minimum value of reflection achieved while retaining optimum microwave absorption. Further the loading of Cu crosses the threshold limit (as observed in the conducting behaviour) and SE_{Tot} decreases, suggesting that Cu₁₅PdCNTs are found to have an optimum composition. However, the EMI SE of pristine MWCNTs is shown in the ESI (Fig. 5(S)†).

The dielectric tangent loss ($\tan \delta \varepsilon = \varepsilon''/\varepsilon'$) of $Cu_x PdCNT$ nanocomposites is shown in Fig. 7(C). The observed $\tan \delta$ values for all the nanocomposites were ≥ 0.2 in the entire frequency range, which is caused by the good conductivity of Cu and Pd nanoparticles and their associated polarization and relaxation. Two hoops are observed for these nanocomposites as revealed in Fig. 7(C), which suggest the excellent microwave shielding performance of $Cu_x PdCNT$ nanocomposites, which mainly may be attributed to two factors: impedance matching and EM wave attenuation.

The EMI shielding using highly conducting materials, such as metals, is known to be governed by reflection rather than absorption. On the other hand, hybrid conducting nanocomposites provide EMI shielding predominantly due to absorption owing to the presence of electric dipoles. 41,48 We observed a similar phenomenon for our nanocomposites, wherein the total EMI SE achieved is greater than the recommended limit of SE (20 dB) for commercial applications. 49

Therefore, the as-fabricated Cu decorated buckypaper ($Cu_{15}PdCNTs$) could be a potential choice for futuristic EMI shielding materials. Interestingly, the desired value of EMI SE can be achieved by scaling the loading level of Cu nanoparticles depending on the specific requirement in the desired application. Our $Cu_{15}PdCNT$ nanocomposites exhibited improved microwave absorption properties in comparison to pristine PdCNTs, MWCNTs, and PANI/MWCNT nanocomposites reported earlier. 50,51

4. Conclusions

We have demonstrated the synthesis of a lightweight and flexible Cu nanoparticle decorated PdCNT buckypaper with high EMI shielding effectiveness, using a simple and cost-effective electroless technique. These nanocomposites have been characterized by employing XRD, FESEM, HRTEM, and Raman spectroscopy, which suggest a uniform selective decoration of Cu nanoparticles on PdCNTs up to a loading limit of \sim 15 wt%, beyond which the Cu nanoparticles tend to agglomerate. The EMI shielding effectiveness of 35 dB (~99.99%) and a high specific EMI SE of ~108 dB cm³ g⁻¹ have been realized in the Ku-band for Cu_xPdCNT buckypaper (thickness of 200 μm) at 15 wt% loading of Cu nanoparticles. It is suggested that the Cu nanoparticle decorated Pd doped MWCNTs could find promising EMI shielding applications in the Ku band, wherein the EMI SE can be tuned depending on the application by tailoring the Cu nanoparticle loading on MWCNTs.

Acknowledgements

The authors are thankful to Mr Naval, Mr Jay Tavale, Dr Sakshi and Ms Kriti for their valuable help in X-ray diffraction and SEM/TEM studies, respectively. One of the authors (Mr Anil Kumar) is thankful to MNRE, India for the financial assistance.

References

- 1 S. Yang, K. Lozano, A. Lomeli, H. D. Foltz and R. Jones, Compos. Appl. Sci. Manuf., 2005, 36, 691–697.
- 2 N. Li, Y. Huang, F. Du, X. He, X. Lin, H. Gao, Y. Ma, F. Li, Y. Chen and P. C. Eklund, *Nano Lett.*, 2006, **6**, 1141–1145.
- 3 M. H. Al-Saleh and U. Sundararaj, *Carbon*, 2009, 47, 1738–1746.
- 4 A. P. Singh, M. Mishra, P. Sambyal, B. K. Gupta, B. P. Singh, A. Chandra and S. K. Dhawan, *J. Mater. Chem. A*, 2014, 2, 3581–3593.
- 5 J. Wang, C. Xiang, Q. Liu, Y. Pan and J. Guo, Adv. Funct. Mater., 2008, 18, 2995–3002.
- 6 Y. Yang, M. C. Gupta, K. L. Dudley and R. W. Lawrence, *Nano Lett.*, 2005, 5, 2131–2134.
- 7 A. Celzard, E. McRae, G. Furdin and J. F. Marêché, *J. Phys.: Condens. Matter*, 1997, **9**, 2225.
- 8 S. H. Park, P. Theilmann, K. Yang, A. M. Rao and P. R. Bandaru, *Appl. Phys. Lett.*, 2010, **96**, 043115.
- 9 S. Palaniappan and B. H. Narayana, *J. Polym. Sci., Part A: Polym. Chem.*, 1994, 32, 2431–2436.

- 10 T. K. Gupta, B. P. Singh, S. R. Dhakate, V. N. Singh and R. B. Mathur, *J. Mater. Chem. A*, 2013, 1, 9138–9149.
- 11 K. Singh, A. Ohlan, V. H. Pham, Balasubramaniyan R., S. Varshney, J. Jang, S. H. Hur, W. M. Choi, M. Kumar, S. K. Dhawan, B.-S. Kong and J. S. Chung, *Nanoscale*, 2013, 5, 2411–2420.
- 12 R. Kumar, S. R. Dhakate, T. Gupta, P. Saini, B. P. Singh and R. B. Mathur, *J. Mater. Chem. A*, 2013, **1**, 5727–5735.
- 13 Q. Liu, J. Gu, W. Zhang, Y. Miyamoto, Z. Chen and D. Zhang, J. Mater. Chem., 2012, 22, 21183–21188.
- 14 M. H. Al-Saleh, W. H. Saadeh and U. Sundararaj, *Carbon*, 2013, **60**, 146–156.
- 15 Y. Chen, Y. Li, M. Yip and N. Tai, *Compos. Sci. Technol.*, 2013, **80**, 80–86.
- 16 S.-S. Tzeng and F.-Y. Chang, Mater. Sci. Eng., A, 2001, 302, 258–267.
- 17 A. A. Al-Ghamdi and F. El-Tantawy, *Compos. Appl. Sci. Manuf.*, 2010, 41, 1693–1701.
- 18 J. Zhaoa and R.-H. Xieb, J. Nanosci. Nanotechnol., 2003, 3, 459–478.
- 19 A. Kumar, M. Kaur, R. Kumar, P. Sengupta, V. Raman, G. Bhatia and K. Sood, J. Mater. Sci., 2010, 45, 1393–1400.
- 20 A. Kumar, A. P. Singh, S. Kumari, P. K. Dutta, S. K. Dhawan and A. Dhar, *J. Mater. Chem. A*, 2014, 2, 16632–16639.
- 21 J. M. Planeix, N. Coustel, B. Coq, V. Brotons, P. S. Kumbhar, R. Dutartre, P. Geneste, P. Bernier and P. M. Ajayan, *J. Am. Chem. Soc.*, 1994, 116, 7935–7936.
- 22 F. Wang, S. Arai and M. Endo, *Electrochem. Commun.*, 2004, 6, 1042–1044.
- 23 A. V. Ellis, K. Vijayamohanan, R. Goswami, N. Chakrapani, L. S. Ramanathan, P. M. Ajayan and G. Ramanath, *Nano Lett.*, 2003, 3, 279–282.
- 24 B. M. Quinn, C. Dekker and S. G. Lemay, *J. Am. Chem. Soc.*, 2005, **127**, 6146–6147.
- 25 Y. Zhang, N. W. Franklin, R. J. Chen and H. Dai, *Chem. Phys. Lett.*, 2000, **331**, 35-41.
- 26 A. P. Singh, B. K. Gupta, M. Mishra, Govind, A. Chandra, R. B. Mathur and S. K. Dhawan, *Carbon*, 2013, **56**, 86–96.
- 27 H. M. Kim, K. Kim, C. Y. Lee, J. Joo, S. J. Cho, H. S. Yoon, D. A. Pejaković, J. W. Yoo and A. J. Epstein, *Appl. Phys. Lett.*, 2004, **84**, 589–591.
- 28 B.-W. Li, Y. Shen, Z.-X. Yue and C.-W. Nan, *Appl. Phys. Lett.*, 2006, **89**, 132504.
- 29 W. Zhao, M. Li, Z. Zhang and H.-X. Peng, *Int. J. Smart Nano Mater.*, 2010, **1**, 249–260.
- 30 P.-G. Ren, D.-X. Yan, X. Ji, T. Chen and Z.-M. Li, *Nanotechnology*, 2011, 22, 055705.

- 31 R. Sharma, R. C. Agarwala and V. Agarwala, *Appl. Surf. Sci.*, 2006, **252**, 8487–8493.
- 32 M. Endo, H. Muramatsu, T. Hayashi, Y. A. Kim, M. Terrones and M. S. Dresselhaus, *Nature*, 2005, 433, 476.
- 33 Z. Wu, Z. Chen, X. Du, J. M. Logan, J. Sippel, M. Nikolou, K. Kamaras, J. R. Reynolds, D. B. Tanner, A. F. Hebard and A. G. Rinzler, *Science*, 2004, **305**, 1273–1276.
- 34 N. Jha and S. Ramaprabhu, J. Phys. Chem. C, 2008, 112, 9315–9319.
- 35 H.-C. Kuan, C.-C. M. Ma, W.-P. Chang, S.-M. Yuen, H.-H. Wu and T.-M. Lee, *Compos. Sci. Technol.*, 2005, **65**, 1703–1710.
- 36 R. Zacharia, K. Y. Kim, A. K. M. Fazle Kibria and K. S. Nahm, Chem. Phys. Lett., 2005, 412, 369–375.
- 37 P. C. Ma, B. Z. Tang and J.-K. Kim, Carbon, 2008, 46, 1497–1505.
- 38 A. M. Nicolson and G. F. Ross, *IEEE Trans. Instrum. Meas.*, 1970, **19**, 377–382.
- 39 Y.-J. Chen, P. Gao, R.-X. Wang, C.-L. Zhu, L.-J. Wang, M.-S. Cao and H.-B. Jin, *J. Phys. Chem. C*, 2009, **113**, 10061– 10064.
- 40 A. P. Singh, M. Mishra, A. Chandra and S. K. Dhawan, Nanotechnology, 2011, 22, 9.
- 41 A. P. Singh, P. Garg, F. Alam, K. Singh, R. B. Mathur, R. P. Tandon, A. Chandra and S. K. Dhawan, *Carbon*, 2012, 50, 3868–3875.
- 42 K. Singh, A. Ohlan, P. Saini and S. K. Dhawan, *Polym. Adv. Technol.*, 2008, **19**, 229–236.
- 43 V. K. Sachdev, K. Patel, S. Bhattacharya and R. P. Tandon, *J. Appl. Polym. Sci.*, 2011, **120**, 1100–1105.
- 44 A. P. Singh, M. Mishra, P. Sambyal, B. K. Gupta, B. P. Singh, A. Chandra and S. K. Dhawan, *J. Mater. Chem. A*, 2014, 2, 3581–3593.
- M. Ashokkumar, N. T. Narayanan, B. K. Gupta, A. L. M. Reddy,
 A. P. Singh, S. K. Dhawan, B. Chandrasekaran, D. Rawat,
 S. Talapatra, P. M. Ajayan and P. Thanikaivelan, ACS Sustainable Chem. Eng., 2013, 1, 619–626.
- 46 N. C. Das, D. Das, T. K. Khastgir and A. C. Chakrraborthy, *Compos. Appl. Sci. Manuf.*, 2000, **31**, 1069–1081.
- 47 N. F. Colaneri and L. W. Shacklette, *IEEE Trans. Instrum. Meas.*, 1992, **41**, 29.
- 48 R. Che, L. M. Peng, X. Duan, Q. Chen and X. Liang, *Adv. Mater.*, 2004, **16**, 401–405.
- 49 N. Li, Y. Huang, F. Du, X. He, X. Lin, H. Gao, Y. Ma, F. Li, Y. Chen and P. C. Eklund, *Nano Lett.*, 2006, **6**, 1141–1145.
- 50 A. P. Singh, A. Kumar, A. Chandra and S. K. Dhawan, *AIP Adv.*, 2011, 1, 022147.
- 51 K. Singh, A. Ohlan, R. K. Kotnala, A. K. Bakhshi and S. K. Dhawan, *Mater. Chem. Phys.*, 2008, **112**, 651–658.